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In alloys. however. composition is also variable and to understand phase changes in ' alloys requires an appreciation of how the Gibbs free energy of a given phase : depends on composition as well as temperature and pressure Since the phase I I G . transformations described in this book mainly occur at a fixed pressure of . I 1 atm most attention will be given to changes in composition and

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The great majority of phase transformations in metals occur by the growth of a new phase (P) from a few nucleation sites within the parent phase (a)-a nucleation and growth process. The a / P interface therefore plays an important role in determining the kinetics of phase transformations and is the most important class of interface listed. It is. however, also the most complex and least

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It is also found to apply to a wide range of processes and transformations in metals and alloys, the simplest of these is the process of diffusion which is discussed in Chapter 2. Exercises 1.1 The

specific heat of solid copper above 300 K References 1. See D .R. Gaskell, Introduciron to Metallurgicai Thermodynamics, p. 342, McGraw-Hill, 1973. 2. For a more detailed treatment of the structures

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